

4-*tert*-Butyl-2-[[*(E)*-2,6-diisopropylphenylimino]methyl]-6-(1-phenylethyl)phenol

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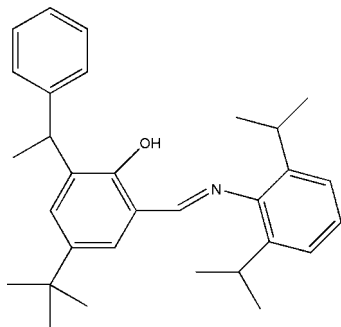
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Key indicators: single-crystal X-ray study; $T = 294$ K; mean $\sigma(\text{C}-\text{C}) = 0.005$ Å; disorder in main residue; R factor = 0.038; wR factor = 0.105; data-to-parameter ratio = 7.9.

The title compound, $\text{C}_{31}\text{H}_{39}\text{NO}$, was prepared by the condensation of 5-*tert*-butyl-2-hydroxy-3-(1-phenylethyl)benzaldehyde with 2,6-diisopropylphenylamine in refluxing ethanol. The phenol ring and the phenylimino ring are approximately perpendicular, forming a dihedral angle of $81.5(3)^\circ$. The phenyl ring makes a dihedral angle of $75.2(2)^\circ$ with the phenol ring and also makes a dihedral angle of $62.9(3)^\circ$ with the phenylimino ring. The molecular structure is stabilized by an intramolecular $\text{O}-\text{H}\cdots\text{N}$ hydrogen bond linking the phenol OH group and the phenylimino N atom. One of the isopropyl groups is disordered over two positions; the site occupancies are 0.55 and 0.45.

Related literature

For related literature, see: Gibson & Spitzmesser (2003); Ittel & Johnson (2000); Shriner *et al.* (1946).



Experimental

Crystal data

$\text{C}_{31}\text{H}_{39}\text{NO}$
 $M_r = 441.63$
 Orthorhombic, $Pca2_1$
 $a = 25.273(4)$ Å
 $b = 9.5496(18)$ Å
 $c = 11.115(2)$ Å

$V = 2682.6(8)$ Å³
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.07$ mm⁻¹
 $T = 294(2)$ K
 $0.24 \times 0.20 \times 0.16$ mm

Data collection

Bruker SMART APEX CCD diffractometer
 Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
 $T_{\min} = 0.974$, $T_{\max} = 0.990$

13332 measured reflections
 2505 independent reflections
 1964 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.039$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.038$
 $wR(F^2) = 0.105$
 $S = 1.06$
 2505 reflections
 319 parameters
 41 restraints

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\max} = 0.13$ e Å⁻³
 $\Delta\rho_{\min} = -0.12$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{O1}-\text{H1}\cdots\text{N1}$	0.93 (4)	1.79 (4)	2.620 (3)	147 (3)

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINTE* (Bruker, 1997); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CS2047).

References

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supplementary materials

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4-*tert*-Butyl-2- $\{[(E)$ -2,6-diisopropylphenylimino]methyl}-6-(1-phenylethyl)phenol

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Comment

Late-metal catalysts bearing salicylaldiminato-type ligands play an important role in the catalytic oligomerization and polymerization of olefins (Ittel *et al.*, 2000). Increasing the size of the imino substituent and the *o*-phenoxy substituent has a pronounced effect on the activity of the catalyst and the molecular weight of the polymer (Gibson *et al.*, 2003). We have recently prepared a novel ligand, (I), and report its crystal structure here.

The molecular structure of (I) is shown in Fig. 1. In (I) the phenyl ring and the phenoxy ring form a dihedral angle of 75.18 (1)°, the phenylimino ring is nearly perpendicular to the phenoxy ring, with a dihedral angle of 81.51 (2)°. The dihedral angle between the phenyl and the phenylimino ring is 62.88 (1)°. Three C atoms of isopropyl attaching to the phenylimino ring are disordered in two locations with 55% and 45% probabilities. An intramolecular hydrogen bond is formed between the hydroxyl group and the N atom stabilizing the crystal structure.

Experimental

2,6-Diisopropylaniline (3.94 g, 22.2 mmol) was added *via* syringe to a solution of 5-*tert*-Butyl-2-hydroxy-3-(1-phenylethyl)-benzaldehyde (6.26 g, 22.2 mmol) in EtOH (100 ml). Formic acid (0.01 g, 0.22 mmol) was added and the solution refluxed for 20 h. The solution was dried over magnesium sulfate, filtered and the volatiles removed under reduced pressure. Extraction into pentane (5 ml) followed by cooling to 243 K afforded yellow crystals of (I) (yield 76%). Spectroscopic analysis: IR (KBr, ν , cm^{-1}): 3055.8, 3027.2, 2965.6, 2871.9, 1621.5, 1591.1, 1460.4, 1361.9, 1269.7, 1176.9, 855.7; ^1H NMR (CDCl_3 , δ , p.p.m.): 1.188 (s, 12H), 1.305 (s, 9H), 1.692 (s, 3H), 3.021 (s, 2H), 4.756 (s, 1H), 7.182–7.369 (m, 10H), 8.287 (s, 1H), 13.146 (s, 1H)

Refinement

Except the hydroxyl H atom, all H atoms were positioned geometrically ($\text{C—H}=0.93\text{--}0.97\text{ \AA}$) and refined as riding. $U_{\text{iso}}(\text{H})$ values are equal to $1.2U_{\text{eq}}(\text{carrier atom})$. In absence of significant anomalous dispersion effects, Friedel-pair reflections were merged prior to refinement.

Figures

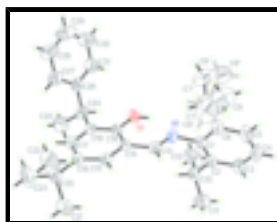


Fig. 1. View of the molecule of (I) showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 35% probability level.

4-*tert*-Butyl-2-[[*(E)*-2,6-diisopropylphenylimino]methyl]-6-(1-phenylethyl)phenol

Crystal data

$C_{31}H_{39}NO$	$D_x = 1.093 \text{ Mg m}^{-3}$
$M_r = 441.63$	Melting point: 400 K
Orthorhombic, $Pca2_1$	Mo $K\alpha$ radiation
$a = 25.273 (4) \text{ \AA}$	$\lambda = 0.71073 \text{ \AA}$
$b = 9.5496 (18) \text{ \AA}$	Cell parameters from 3934 reflections
$c = 11.115 (2) \text{ \AA}$	$\theta = 2.3\text{--}24.1^\circ$
$V = 2682.6 (8) \text{ \AA}^3$	$\mu = 0.07 \text{ mm}^{-1}$
$Z = 4$	$T = 294 (2) \text{ K}$
$F_{000} = 960$	Block, yellow
	$0.24 \times 0.20 \times 0.16 \text{ mm}$

Data collection

Bruker SMART APEX CCD diffractometer	2505 independent reflections
Radiation source: fine-focus sealed tube	1964 reflections with $I > 2\sigma(I)$
Monochromator: graphite	$R_{\text{int}} = 0.039$
$T = 294(2) \text{ K}$	$\theta_{\text{max}} = 25.0^\circ$
φ and ω scans	$\theta_{\text{min}} = 2.1^\circ$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$h = -19 \rightarrow 30$
$T_{\text{min}} = 0.974$, $T_{\text{max}} = 0.990$	$k = -11 \rightarrow 9$
13332 measured reflections	$l = -11 \rightarrow 13$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.038$	H atoms treated by a mixture of independent and constrained refinement
$wR(F^2) = 0.105$	$w = 1/[\sigma^2(F_o^2) + (0.0512P)^2 + 0.4436P]$
$S = 1.06$	where $P = (F_o^2 + 2F_c^2)/3$
2505 reflections	$(\Delta/\sigma)_{\text{max}} = 0.003$
319 parameters	$\Delta\rho_{\text{max}} = 0.13 \text{ e \AA}^{-3}$
41 restraints	$\Delta\rho_{\text{min}} = -0.12 \text{ e \AA}^{-3}$
Primary atom site location: structure-invariant direct methods	Extinction correction: none

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
O1	0.44217 (8)	0.0843 (2)	0.9096 (2)	0.0504 (5)	
H1	0.4093 (15)	0.115 (4)	0.937 (4)	0.076*	
N1	0.34993 (8)	0.2094 (3)	0.9030 (2)	0.0441 (6)	
C1	0.29497 (10)	0.2202 (3)	0.9371 (3)	0.0420 (7)	
C2	0.26067 (11)	0.1171 (3)	0.8930 (3)	0.0456 (7)	
C3	0.20812 (12)	0.1214 (4)	0.9313 (3)	0.0571 (9)	
H3A	0.1845	0.0538	0.9039	0.069*	
C4	0.19064 (13)	0.2234 (4)	1.0087 (4)	0.0647 (10)	
H4A	0.1555	0.2246	1.0331	0.078*	
C5	0.22512 (13)	0.3240 (4)	1.0501 (3)	0.0597 (9)	
H5A	0.2126	0.3936	1.1012	0.072*	
C6	0.27828 (12)	0.3240 (3)	1.0173 (3)	0.0489 (8)	
C7	0.31638 (13)	0.4301 (4)	1.0708 (3)	0.0585 (9)	0.55
H7A	0.3432	0.4506	1.0096	0.070*	0.55
C8	0.3442 (4)	0.3728 (10)	1.1771 (9)	0.093 (3)	0.55
H8A	0.3612	0.2864	1.1558	0.140*	0.55
H8B	0.3192	0.3560	1.2404	0.140*	0.55
H8C	0.3703	0.4387	1.2042	0.140*	0.55
C9	0.2903 (4)	0.5695 (8)	1.1056 (11)	0.094 (3)	0.55
H9A	0.2724	0.6078	1.0370	0.141*	0.55
H9B	0.3169	0.6340	1.1324	0.141*	0.55
H9C	0.2653	0.5539	1.1692	0.141*	0.55
C7'	0.31638 (13)	0.4301 (4)	1.0708 (3)	0.0585 (9)	0.45
H7'A	0.3517	0.4052	1.0413	0.070*	0.45
C8'	0.3184 (5)	0.4162 (15)	1.2099 (8)	0.101 (4)	0.45
H8'A	0.3426	0.4838	1.2420	0.152*	0.45
H8'B	0.3299	0.3237	1.2311	0.152*	0.45
H8'C	0.2837	0.4325	1.2426	0.152*	0.45
C9'	0.3053 (5)	0.5723 (10)	1.0256 (14)	0.101 (4)	0.45
H9'A	0.3053	0.5715	0.9392	0.152*	0.45
H9'B	0.3320	0.6356	1.0540	0.152*	0.45
H9'C	0.2712	0.6023	1.0540	0.152*	0.45
C10	0.27988 (12)	0.0008 (3)	0.8111 (3)	0.0538 (8)	
H10A	0.3106	0.0365	0.7670	0.065*	
C11	0.23939 (16)	-0.0449 (5)	0.7194 (4)	0.0823 (13)	
H11A	0.2287	0.0343	0.6720	0.124*	
H11B	0.2091	-0.0831	0.7599	0.124*	
H11C	0.2546	-0.1149	0.6679	0.124*	
C12	0.29860 (19)	-0.1216 (4)	0.8871 (4)	0.0941 (14)	
H12A	0.3245	-0.0896	0.9440	0.141*	
H12B	0.3141	-0.1915	0.8360	0.141*	
H12C	0.2690	-0.1613	0.9293	0.141*	
C13	0.36706 (10)	0.2802 (3)	0.8142 (3)	0.0403 (6)	
H13A	0.3447	0.3457	0.7788	0.048*	
C14	0.42028 (10)	0.2630 (3)	0.7657 (2)	0.0344 (6)	

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C15	0.43582 (11)	0.3420 (3)	0.6666 (3)	0.0390 (6)
H15A	0.4126	0.4092	0.6367	0.047*
C16	0.48415 (11)	0.3248 (3)	0.6105 (3)	0.0394 (7)
C17	0.51775 (11)	0.2242 (3)	0.6607 (3)	0.0415 (7)
H17A	0.5506	0.2102	0.6251	0.050*
C18	0.50481 (10)	0.1442 (3)	0.7606 (3)	0.0378 (6)
C19	0.45548 (10)	0.1634 (3)	0.8132 (2)	0.0351 (6)
C20	0.50008 (12)	0.4134 (3)	0.5021 (3)	0.0501 (8)
C21	0.45576 (16)	0.4140 (5)	0.4080 (4)	0.0873 (13)
H21A	0.4237	0.4480	0.4440	0.131*
H21B	0.4502	0.3205	0.3788	0.131*
H21C	0.4655	0.4738	0.3423	0.131*
C22	0.50891 (19)	0.5640 (4)	0.5434 (4)	0.0891 (14)
H22A	0.4775	0.5980	0.5823	0.134*
H22B	0.5168	0.6218	0.4751	0.134*
H22C	0.5380	0.5669	0.5990	0.134*
C23	0.55002 (16)	0.3593 (5)	0.4403 (4)	0.0898 (14)
H23A	0.5787	0.3581	0.4969	0.135*
H23B	0.5589	0.4197	0.3742	0.135*
H23C	0.5439	0.2662	0.4109	0.135*
C24	0.54360 (11)	0.0404 (3)	0.8163 (3)	0.0484 (7)
H24A	0.5223	-0.0311	0.8570	0.058*
C25	0.57752 (15)	-0.0359 (4)	0.7233 (4)	0.0720 (11)
H25A	0.6009	-0.0999	0.7635	0.108*
H25B	0.5980	0.0310	0.6787	0.108*
H25C	0.5550	-0.0869	0.6692	0.108*
C26	0.57654 (11)	0.1120 (3)	0.9126 (3)	0.0453 (7)
C27	0.61133 (12)	0.2191 (4)	0.8845 (3)	0.0595 (9)
H27A	0.6149	0.2469	0.8047	0.071*
C28	0.64090 (15)	0.2858 (4)	0.9719 (4)	0.0753 (12)
H28A	0.6639	0.3579	0.9511	0.090*
C29	0.63612 (18)	0.2449 (5)	1.0891 (4)	0.0883 (14)
H29A	0.6560	0.2889	1.1485	0.106*
C30	0.60198 (18)	0.1388 (5)	1.1194 (4)	0.0839 (12)
H30A	0.5987	0.1112	1.1993	0.101*
C31	0.57252 (13)	0.0728 (4)	1.0313 (3)	0.0631 (9)
H31A	0.5496	0.0007	1.0526	0.076*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0451 (11)	0.0618 (13)	0.0441 (12)	0.0068 (10)	0.0068 (10)	0.0183 (11)
N1	0.0306 (11)	0.0593 (15)	0.0424 (14)	-0.0006 (11)	0.0055 (11)	0.0033 (13)
C1	0.0327 (14)	0.0546 (17)	0.0388 (16)	0.0000 (13)	0.0058 (12)	0.0033 (15)
C2	0.0379 (16)	0.0617 (18)	0.0373 (16)	-0.0015 (13)	0.0073 (13)	0.0017 (15)
C3	0.0375 (16)	0.079 (2)	0.055 (2)	-0.0110 (15)	0.0061 (16)	-0.0081 (19)
C4	0.0375 (17)	0.083 (2)	0.073 (2)	-0.0009 (17)	0.0173 (18)	-0.004 (2)
C5	0.0496 (19)	0.068 (2)	0.062 (2)	0.0047 (16)	0.0191 (18)	-0.0052 (19)

C6	0.0430 (17)	0.0568 (19)	0.0468 (18)	-0.0007 (14)	0.0076 (15)	-0.0001 (16)
C7	0.054 (2)	0.058 (2)	0.063 (2)	-0.0073 (16)	0.0126 (17)	-0.0091 (18)
C8	0.088 (6)	0.087 (5)	0.105 (6)	-0.007 (4)	-0.039 (5)	0.009 (5)
C9	0.103 (6)	0.062 (4)	0.117 (7)	0.002 (4)	-0.016 (5)	-0.004 (5)
C7'	0.054 (2)	0.058 (2)	0.063 (2)	-0.0073 (16)	0.0126 (17)	-0.0091 (18)
C8'	0.097 (7)	0.107 (8)	0.100 (7)	-0.017 (6)	-0.023 (6)	-0.001 (6)
C9'	0.104 (7)	0.076 (6)	0.123 (8)	-0.008 (5)	0.000 (7)	0.017 (6)
C10	0.0483 (17)	0.071 (2)	0.0425 (17)	-0.0054 (15)	0.0089 (16)	-0.0042 (18)
C11	0.078 (3)	0.102 (3)	0.067 (2)	-0.020 (2)	0.001 (2)	-0.028 (2)
C12	0.129 (4)	0.083 (3)	0.070 (3)	0.034 (3)	0.023 (3)	-0.001 (3)
C13	0.0318 (14)	0.0524 (16)	0.0368 (15)	0.0004 (12)	-0.0002 (13)	0.0025 (15)
C14	0.0303 (14)	0.0432 (14)	0.0298 (13)	-0.0047 (11)	-0.0031 (11)	0.0008 (13)
C15	0.0345 (15)	0.0472 (15)	0.0351 (15)	0.0012 (12)	-0.0038 (13)	0.0037 (14)
C16	0.0347 (15)	0.0487 (16)	0.0346 (15)	-0.0063 (12)	-0.0004 (13)	0.0021 (13)
C17	0.0293 (14)	0.0566 (17)	0.0385 (15)	-0.0018 (12)	0.0038 (12)	0.0012 (15)
C18	0.0327 (15)	0.0434 (15)	0.0371 (16)	0.0001 (11)	-0.0028 (12)	0.0002 (14)
C19	0.0337 (14)	0.0413 (14)	0.0303 (14)	-0.0061 (11)	-0.0007 (12)	0.0023 (13)
C20	0.0473 (18)	0.062 (2)	0.0412 (18)	-0.0058 (14)	0.0078 (14)	0.0118 (16)
C21	0.081 (3)	0.131 (4)	0.051 (2)	-0.006 (2)	-0.005 (2)	0.026 (3)
C22	0.116 (4)	0.070 (3)	0.081 (3)	-0.027 (2)	0.019 (3)	0.021 (2)
C23	0.073 (3)	0.116 (3)	0.081 (3)	0.012 (2)	0.038 (2)	0.038 (3)
C24	0.0387 (16)	0.0498 (17)	0.0567 (19)	0.0062 (13)	0.0012 (15)	0.0023 (16)
C25	0.067 (2)	0.077 (3)	0.072 (2)	0.024 (2)	-0.003 (2)	-0.007 (2)
C26	0.0308 (14)	0.0571 (18)	0.0479 (18)	0.0103 (13)	0.0022 (14)	0.0090 (16)
C27	0.0482 (19)	0.073 (2)	0.057 (2)	0.0023 (17)	-0.0059 (17)	0.0125 (19)
C28	0.055 (2)	0.080 (3)	0.091 (3)	-0.0040 (18)	-0.025 (2)	0.013 (2)
C29	0.082 (3)	0.099 (3)	0.084 (3)	0.009 (3)	-0.044 (3)	0.000 (3)
C30	0.089 (3)	0.106 (3)	0.057 (2)	0.011 (3)	-0.021 (2)	0.018 (2)
C31	0.057 (2)	0.079 (2)	0.054 (2)	0.0035 (17)	-0.0029 (18)	0.0169 (19)

Geometric parameters (Å, °)

O1—C19	1.354 (3)	C13—H13A	0.9300
O1—H1	0.93 (4)	C14—C15	1.392 (4)
N1—C13	1.272 (4)	C14—C19	1.405 (4)
N1—C1	1.443 (3)	C15—C16	1.381 (4)
C1—C6	1.398 (4)	C15—H15A	0.9300
C1—C2	1.400 (4)	C16—C17	1.398 (4)
C2—C3	1.395 (4)	C16—C20	1.526 (4)
C2—C10	1.516 (4)	C17—C18	1.387 (4)
C3—C4	1.372 (5)	C17—H17A	0.9300
C3—H3A	0.9300	C18—C19	1.389 (4)
C4—C5	1.376 (5)	C18—C24	1.525 (4)
C4—H4A	0.9300	C20—C22	1.527 (5)
C5—C6	1.392 (4)	C20—C23	1.527 (5)
C5—H5A	0.9300	C20—C21	1.532 (5)
C6—C7	1.520 (5)	C21—H21A	0.9600
C7—C8	1.480 (7)	C21—H21B	0.9600
C7—C9	1.535 (7)	C21—H21C	0.9600

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C7—H7A	0.9800	C22—H22A	0.9600
C8—H8A	0.9600	C22—H22B	0.9600
C8—H8B	0.9600	C22—H22C	0.9600
C8—H8C	0.9600	C23—H23A	0.9600
C9—H9A	0.9600	C23—H23B	0.9600
C9—H9B	0.9600	C23—H23C	0.9600
C9—H9C	0.9600	C24—C26	1.519 (5)
C8'—H8'A	0.9600	C24—C25	1.528 (5)
C8'—H8'B	0.9600	C24—H24A	0.9800
C8'—H8'C	0.9600	C25—H25A	0.9600
C9'—H9'A	0.9600	C25—H25B	0.9600
C9'—H9'B	0.9600	C25—H25C	0.9600
C9'—H9'C	0.9600	C26—C31	1.375 (5)
C10—C11	1.509 (5)	C26—C27	1.385 (4)
C10—C12	1.517 (5)	C27—C28	1.381 (5)
C10—H10A	0.9800	C27—H27A	0.9300
C11—H11A	0.9600	C28—C29	1.366 (6)
C11—H11B	0.9600	C28—H28A	0.9300
C11—H11C	0.9600	C29—C30	1.373 (6)
C12—H12A	0.9600	C29—H29A	0.9300
C12—H12B	0.9600	C30—C31	1.383 (6)
C12—H12C	0.9600	C30—H30A	0.9300
C13—C14	1.458 (4)	C31—H31A	0.9300
C19—O1—H1	108 (2)	C16—C15—H15A	118.6
C13—N1—C1	119.5 (2)	C14—C15—H15A	118.6
C6—C1—C2	122.3 (3)	C15—C16—C17	116.0 (3)
C6—C1—N1	120.6 (3)	C15—C16—C20	121.6 (3)
C2—C1—N1	117.0 (3)	C17—C16—C20	122.4 (2)
C3—C2—C1	117.5 (3)	C18—C17—C16	123.7 (3)
C3—C2—C10	120.6 (3)	C18—C17—H17A	118.1
C1—C2—C10	121.8 (2)	C16—C17—H17A	118.1
C4—C3—C2	121.2 (3)	C17—C18—C19	118.4 (2)
C4—C3—H3A	119.4	C17—C18—C24	122.1 (3)
C2—C3—H3A	119.4	C19—C18—C24	119.5 (3)
C3—C4—C5	120.1 (3)	O1—C19—C18	118.8 (2)
C3—C4—H4A	120.0	O1—C19—C14	121.2 (2)
C5—C4—H4A	120.0	C18—C19—C14	120.0 (2)
C4—C5—C6	121.6 (3)	C16—C20—C22	108.8 (3)
C4—C5—H5A	119.2	C16—C20—C23	112.7 (3)
C6—C5—H5A	119.2	C22—C20—C23	109.5 (3)
C5—C6—C1	117.3 (3)	C16—C20—C21	110.4 (3)
C5—C6—C7	120.6 (3)	C22—C20—C21	108.0 (3)
C1—C6—C7	122.1 (3)	C23—C20—C21	107.4 (3)
C8—C7—C6	111.5 (5)	C20—C21—H21A	109.5
C8—C7—C9	108.9 (7)	C20—C21—H21B	109.5
C6—C7—C9	113.9 (4)	H21A—C21—H21B	109.5
C8—C7—H7A	107.4	C20—C21—H21C	109.5
C6—C7—H7A	107.4	H21A—C21—H21C	109.5
C9—C7—H7A	107.4	H21B—C21—H21C	109.5

C7—C8—H8A	109.5	C20—C22—H22A	109.5
C7—C8—H8B	109.5	C20—C22—H22B	109.5
H8A—C8—H8B	109.5	H22A—C22—H22B	109.5
C7—C8—H8C	109.5	C20—C22—H22C	109.5
H8A—C8—H8C	109.5	H22A—C22—H22C	109.5
H8B—C8—H8C	109.5	H22B—C22—H22C	109.5
C7—C9—H9A	109.5	C20—C23—H23A	109.5
C7—C9—H9B	109.5	C20—C23—H23B	109.5
H9A—C9—H9B	109.5	H23A—C23—H23B	109.5
C7—C9—H9C	109.5	C20—C23—H23C	109.5
H9A—C9—H9C	109.5	H23A—C23—H23C	109.5
H9B—C9—H9C	109.5	H23B—C23—H23C	109.5
H8'A—C8'—H8'B	109.5	C26—C24—C18	110.2 (2)
H8'A—C8'—H8'C	109.5	C26—C24—C25	112.6 (3)
H8'B—C8'—H8'C	109.5	C18—C24—C25	113.4 (3)
H9'A—C9'—H9'B	109.5	C26—C24—H24A	106.7
H9'A—C9'—H9'C	109.5	C18—C24—H24A	106.7
H9'B—C9'—H9'C	109.5	C25—C24—H24A	106.7
C11—C10—C2	113.6 (3)	C24—C25—H25A	109.5
C11—C10—C12	111.4 (3)	C24—C25—H25B	109.5
C2—C10—C12	109.3 (3)	H25A—C25—H25B	109.5
C11—C10—H10A	107.4	C24—C25—H25C	109.5
C2—C10—H10A	107.4	H25A—C25—H25C	109.5
C12—C10—H10A	107.4	H25B—C25—H25C	109.5
C10—C11—H11A	109.5	C31—C26—C27	117.7 (3)
C10—C11—H11B	109.5	C31—C26—C24	120.9 (3)
H11A—C11—H11B	109.5	C27—C26—C24	121.4 (3)
C10—C11—H11C	109.5	C28—C27—C26	121.7 (3)
H11A—C11—H11C	109.5	C28—C27—H27A	119.1
H11B—C11—H11C	109.5	C26—C27—H27A	119.1
C10—C12—H12A	109.5	C29—C28—C27	119.4 (4)
C10—C12—H12B	109.5	C29—C28—H28A	120.3
H12A—C12—H12B	109.5	C27—C28—H28A	120.3
C10—C12—H12C	109.5	C28—C29—C30	120.0 (4)
H12A—C12—H12C	109.5	C28—C29—H29A	120.0
H12B—C12—H12C	109.5	C30—C29—H29A	120.0
N1—C13—C14	122.7 (3)	C29—C30—C31	120.1 (4)
N1—C13—H13A	118.6	C29—C30—H30A	120.0
C14—C13—H13A	118.6	C31—C30—H30A	120.0
C15—C14—C19	119.0 (2)	C26—C31—C30	121.1 (4)
C15—C14—C13	119.5 (2)	C26—C31—H31A	119.5
C19—C14—C13	121.4 (2)	C30—C31—H31A	119.5
C16—C15—C14	122.9 (3)		
C13—N1—C1—C6	-87.7 (3)	C16—C17—C18—C19	0.9 (4)
C13—N1—C1—C2	96.7 (3)	C16—C17—C18—C24	-177.1 (3)
C6—C1—C2—C3	0.5 (4)	C17—C18—C19—O1	179.0 (2)
N1—C1—C2—C3	176.1 (3)	C24—C18—C19—O1	-3.0 (4)
C6—C1—C2—C10	-176.7 (3)	C17—C18—C19—C14	-0.7 (4)
N1—C1—C2—C10	-1.2 (4)	C24—C18—C19—C14	177.3 (3)

supplementary materials

C1—C2—C3—C4	0.4 (5)	C15—C14—C19—O1	179.9 (2)
C10—C2—C3—C4	177.8 (3)	C13—C14—C19—O1	-3.3 (4)
C2—C3—C4—C5	-0.1 (6)	C15—C14—C19—C18	-0.4 (4)
C3—C4—C5—C6	-1.3 (6)	C13—C14—C19—C18	176.4 (3)
C4—C5—C6—C1	2.2 (5)	C15—C16—C20—C22	69.1 (4)
C4—C5—C6—C7	-175.8 (3)	C17—C16—C20—C22	-109.4 (3)
C2—C1—C6—C5	-1.8 (4)	C15—C16—C20—C23	-169.3 (3)
N1—C1—C6—C5	-177.2 (3)	C17—C16—C20—C23	12.2 (4)
C2—C1—C6—C7	176.2 (3)	C15—C16—C20—C21	-49.2 (4)
N1—C1—C6—C7	0.8 (4)	C17—C16—C20—C21	132.3 (3)
C5—C6—C7—C8	94.5 (6)	C17—C18—C24—C26	90.3 (3)
C1—C6—C7—C8	-83.4 (6)	C19—C18—C24—C26	-87.6 (3)
C5—C6—C7—C9	-29.2 (7)	C17—C18—C24—C25	-37.0 (4)
C1—C6—C7—C9	152.9 (6)	C19—C18—C24—C25	145.1 (3)
C3—C2—C10—C11	36.4 (4)	C18—C24—C26—C31	115.0 (3)
C1—C2—C10—C11	-146.4 (3)	C25—C24—C26—C31	-117.3 (4)
C3—C2—C10—C12	-88.6 (4)	C18—C24—C26—C27	-64.6 (4)
C1—C2—C10—C12	88.6 (4)	C25—C24—C26—C27	63.1 (4)
C1—N1—C13—C14	-172.8 (3)	C31—C26—C27—C28	-0.5 (5)
N1—C13—C14—C15	178.5 (3)	C24—C26—C27—C28	179.2 (3)
N1—C13—C14—C19	1.7 (4)	C26—C27—C28—C29	0.4 (6)
C19—C14—C15—C16	1.4 (4)	C27—C28—C29—C30	-0.3 (7)
C13—C14—C15—C16	-175.5 (3)	C28—C29—C30—C31	0.2 (7)
C14—C15—C16—C17	-1.2 (4)	C27—C26—C31—C30	0.4 (5)
C14—C15—C16—C20	-179.8 (3)	C24—C26—C31—C30	-179.2 (3)
C15—C16—C17—C18	0.1 (4)	C29—C30—C31—C26	-0.3 (6)
C20—C16—C17—C18	178.7 (3)		

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O1—H1 \cdots N1	0.93 (4)	1.79 (4)	2.620 (3)	147 (3)

Fig. 1

